

# Fundamental Physical Constants / Naturkonstanten

**Note: This page requires a browser capable of math mode, e.g., Arena**

- Version of this page not using math mode / Version ohne "math mode"
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Exponent (base 10) of decimal numbers:  $e\ n = 10^n$  .

Units are separated by spaces, positive numbers denote powers, all units to the right of the slash are in the denominator (negative powers).

**Planck constant** / Plancksches Wirkungsquantum  $h$ :

$$6.6260755e-34\ \text{J s}$$

$$\frac{h}{2\pi} = 1.05457266e-34\ \text{J s}$$

**Boltzmann constant** / Boltzmannsche Konstante

$$k_B : 1.380658e-23\ \text{J/K}$$

**Elementary charge** / Elementarladung  $e$ :

$$1.60217733e-19\ \text{C}$$

**Avogadro number** / Avogadrosche Zahl

$$N_A : 6.0221367e23\ \text{particles/mol}$$

**Speed of light** / Lichtgeschwindigkeit  $c$ :

$$2.99792458e8\ \text{m/s}$$

**Permeability of vacuum** / Permeabilität des Vakuums

$$\mu_0 = 4\pi\ e-7\ \text{T}^2\text{m}^3/\text{J}$$

$$12.566370614e-7\ \text{T}^2\text{m}^3/\text{J}$$

**Permittivity of vacuum** / Dielektrizitätskonstante

$$\varepsilon_0 = \frac{1}{\mu_0 c^2}$$

$$8.854187817e-12\ \text{C}^2/\text{Jm}$$

**Fine structure constant** / Feinstrukturkonstante

$$1 / \alpha = 137.0359895$$

**Electron rest mass** / Elektronenruhemasse

$$m_e : 9.1093897e-31\ \text{kg}$$

**Proton rest mass** / Protonenruhemasse

$$m_p : 1.6726231e-27\ \text{kg}$$

**Neutron rest mass** / Neutronenruhemasse

$$m_n : 1.6749286e-27\ \text{kg}$$

**Bohr magneton** / Bohrsches Magneton

$$\mu_B = \frac{eh}{4\pi m_e}$$

$$9.2740154e-24\ \text{J/T}$$

**Nuclear magneton / Kernmagneton**

$$\mu_N = \frac{eh}{4\pi m_p}$$

$$5.0507866e-27 \text{ J/T}$$

**Free electron g factor / g-Faktor des freien Elektrons**

$$g_e = 2.002319304386$$

**Free electron gyromagnetic ratio / gyromagnetisches Verhältnis des freien Elektrons**

$$\gamma_e = \frac{2\pi g_e \mu_B}{h}$$

$$1.7608592e11 \text{ 1/s T}$$

$$\frac{\gamma_e}{2\pi} = 28.024944 \text{ GHz/T}$$

**Electron magnetic moment / magnetisches Moment des Elektrons**

$$\mu_e = -\frac{1}{2} g_e \mu_B$$

$$-9.2847701e-24 \text{ J/T}$$

**Proton gyromagnetic ratio (H2O) / gyromagnetisches Verhältnis der Protonen in Wasser**

$$\gamma_p = 2.67515255e8 \text{ 1/s T}$$

$$\frac{\gamma_p}{2\pi} = 42.576375 \text{ MHz/T}$$

**Proton magnetic moment / magnetisches Moment des Protons**

$$\mu_p = 1.41060761e-26 \text{ J/T}$$

**Proton-electron ratios / Proton-Elektron-Verhältnisse**

$$\frac{m_p}{m_e} = 1836.152701$$

$$\frac{\mu_e}{\mu_p} = 658.2106881$$

$$\frac{\gamma_e}{\gamma_p} = 658.2275841 \text{ (protons in water)}$$

**Charge-to-mass ratio for electron / Ladungs-Masse-Verhältnis des Elektrons**

$$\frac{e}{m_e} = 1.75880e11 \text{ C/kg}$$

**Atomic mass unit / Einheit der Atommasse amu:**

$$1.66057e-27 \text{ kg}$$

**Bohr radius / Bohrscher Radius**

$$a_0 : 5.29177e-11 \text{ m}$$

**Electron radius / Elektronenradius**

$$r_e : 2.81792e-15 \text{ m}$$

**Gas constant / Gaskonstante R:**

$$R = N_A k_B$$

$$8.31451 \text{ m}^2 \text{ kg/s}^2 \text{ K mol}$$

**Molar volume / Molvolumen**

$$V_{mol} : 22.41383 \text{ m}^3 / \text{kmol}$$

**Faraday constant / Faradaysche Konstante F:**

$$F = N_A e$$

$$9.64846e4 \text{ C/mol}$$

**Proton g factor (Lande factor) / g-Faktor des Protons**

$$g_H : 5.585$$

**Gravitational constant** / Gravitationskonstante  $G$

$6.672\text{e-}11 \text{ m}^3/\text{kg s}^2$  (note: there is controversy about the accurate value of  $G$ )

$6.6732\text{e-}11 \text{ m}^3/\text{kg s}^2$  (from: BRUKER Almanac 1988, possibly inaccurate)

**Acceleration due to gravity** / Erdbeschleunigung  $g$ :

$9.80665 \text{ m/s}^2$

**Compton wavelength of the electron** / Compton-Wellenlänge des Elektrons

$$\lambda_c = \frac{h}{m_e c}$$

$2.42631\text{e-}12 \text{ m}$

## Further Useful Constants / Weitere nützliche Konstanten

**Atomic energy unit** / atomare Energieeinheit **Hartree**

$$1 \text{ Hartree} = \frac{e^2}{4\pi\epsilon_0 a_0}$$

1 Hartree =  $2.625501\text{e+}06 \text{ J/mol}$  (approx. 627.5 kcal/mol)

## Useful Conversion Factors / Nützliche Umrechnungsfaktoren

### NMR (Nuclear Magnetic Resonance)

Proton Larmor frequency

$$\nu_p = \frac{\gamma_p}{2\pi} B$$

$$\nu_p = 42.5764 \text{ MHz/T (H}_2\text{O)}$$

### EPR (Electron Paramagnetic Resonance, ESR)

Electron Larmor frequency

$$\nu_e = \frac{\gamma_e}{2\pi} \frac{g}{g_e} B$$

$$\nu_e / \text{GHz} = 13.9962 g B / \text{T}$$

$$g = 0.07144775 \frac{\nu_e / \text{GHz}}{B / \text{T}}$$

$$g = 3.04199 \frac{\nu_e / \text{GHz}}{\nu_p / \text{MHz}}$$

$$B / \text{T} = 0.0234872 \nu_p / \text{MHz}$$

Conversion of Units

$$1 \text{ G} = 0.1 \text{ mT}$$

$$1 \text{ T} = 10 \text{ kG}$$

$$1 \text{ mT} = 10 \text{ G}$$

$$A / \text{MHz} = 2.80249 \frac{g}{g_e} A / \text{G}$$

$$A / \text{MHz} = 28.0249 \frac{g}{g_e} A / \text{mT}$$

$$A / \text{MHz} = 13.9962 g A / \text{mT}$$

$$A / \text{MHz} = 2.99792e4 A / \text{cm}^{-1}$$

$$A / \text{cm}^{-1} = 0.333564e-4 A / \text{MHz}$$

$$A / \text{cm}^{-1} = 4.66863e-4 g A / \text{mT}$$

- The NIST Reference on Constants, Units, and Uncertainty
  - CODATA Internationally recommended values of the Fundamental Physical Constants
- Fundamental constants C Header File, FORTRAN Data File
- Conversion of units / Umrechnung von Einheiten
- Physical Constants and Astronomical Data (*U Wisconsin*)
- Scientific Constants (*UIUC*)
- Index General chemistry / Allgemeine Chemie

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Note: Physical quantities should be typeset in an *italic* font, whereas units should be typeset in a regular (non-italic) font. Due to problems with currently available browsers, this page (or parts of it) may not be rendered according to these rules.

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*Burkhard Kirste, 1996-10-28*